## Chemical and Physical Changes Web Quest

Go to the following link and answer the following questions: <a href="http://www.ric.edu/faculty/ptiskus/chemical/">http://www.ric.edu/faculty/ptiskus/chemical/</a>

1.	The explosion of is an example of chemical change.						
2.	During a change substances are						
	into substances.						
3.	In other words, the	of the substance	the substance				
4.		changes are relatively easy to					
	If only the	of a substance	you have				
		a physical change.					
Co	olor Changes:						
1.	As in the case of autum	n leaves a in in	is a clue to indicate a				
2.	Perhaps you have found	l a half eaten apple that turns change has occurred when food spoils.	The reason is that a				
En	nergy-Absorbed or Releas	sed:					
1.	Another sign of a	change is the	or of				
	energy by an object.	<i>E</i>					
2.		energy to	a chemical change.				
		during chemical changes involved in					
	<del></del>						
4.	Energy can also be	during a chemical change.					
5.	The	that were mentioned above	energy in a form of				
	that yo	u can see.					
Oc	dor Changes:						
1.	When eggs and food	they undergo a chemical					
2.	The change in	is a clue to the chemical change.					
		to avoid eating	food and becoming				
			•				
Pr	oduction of Gases or Soli	ds:					
1.	The formation of a	is a clue to chemical changes.					
		of gas that you observed form when an					
		e of	15 12 5PP CC				
3.	Another	that a chemical change has occurred is the	of a				
-	solid.						
4.		hat separates out of	during a chemical change is				
	called a precipitate.	•					

Click on Precipitate to check out the pictures. Watch the short video.

2.		and		_ are physical cha	inges.	
	The	prod	luced during _		changes, howe	ever
		easily		back i	nto the	
3.	The most thing for you to remember is that in a					
	of a substance does					
					_ the	of a
	substance		change.			
Cł	nemical Reactions:					
1.	Chemical change Chemical reaction	es are called cher	nical			
١.	Chemical reactio	ns involve		different subst	tances.	_
•	The chemical rea	ection		a new substance	with	and
		physi	cal and chemic	cal properties.	_	
<b>1</b> .	Matter is NEVEL	₹	or		in che	mical reactions.
١.	The	of or	ne substance a	re		_ to form a new
	substance.					
١.	The	r	number of part	icles that exist		_ the reaction exist
		the reaction	l <b>.</b>			
n:	//nobel scas be	cit ca/chem00	10/uni+2/2	3 changes htm	n#	
	//nobel.scas.bo What is a physica		10/unit2/2.	3_changes.htm	<u>n#</u>	
1.		al change?		3_changes.htm	<u>n#</u>	
l.	What is a physica	al change?		3_changes.htm	<u>n#</u>	
1.	What is a physica	al change? of a physical cha		3 changes.htm	<u>n#</u>	

Go to the following website and answer the following questions:
http://antoine.frostburg.edu/chem/senese/101/matter/faq/physical-chemical.shtml
1. What new information can you add to your definition of physical change?
2. What new information can you add to your definition of chemical change?
Go to the following website and quiz your knowledge on physical and chemical changes.
http://www.quia.com/quiz/303980.html

Score: \_\_\_\_\_\_%